

# **Conversion of pyrolysis oil droplets under Entrained Flow Gasifier conditions**

F. Hüsing, T. Kolb

#### **Objectives** Challenges Model based description for conversion of pyrolysis oil under Experimental validation of evaporation model and elevated temperature and pressure: reliable characterization of feedstock: Complex mixture of multiphase composition Numerical approach for depicting the single steps of conversion Multistep conversion and formation of solid residue Input data for numerical simulation of technical EFG Adaption of measuring techniques for droplet evaporation **Fuel Characterization Pyrolysis oil** Substitute fuel Selection of real components to characterize Chemical composition: pyrolysis oil mixtures

### **Objectives:**

 Small number of components representing the evaporation behavior of the technical fuel

# Implementation:

- · Identification of components based on the distillation curve of pyrolysis oil
- Selection of type and amount of component by fractional distillation of boiling range and fraction analysis

# Experimental Setup

CO2 N2 N<sub>2</sub>

# Parameters of interest for Model Validation

- Composition of gas phase
- Droplet size and shape over time course
- Droplet surface temperature
- Conversion of pyrolysis oil to gaseous and solid fractions depends on:
  - Reactor temperature
  - Residence time
  - Fuel composition
- Formation of solid residue for characterization

# • Modelling of Multicomponent Evaporation and Solid Formation

- · Large variety of organic liquids
- · High amount of oxygenated compounds

# **Physical Properties:**

- · High-viscous non-Newtonian fluid
- Wide range of boiling points
- · High amount of non-vaporizing components

# Challenges:

- No complete dissolution of chemical components
- No fractional distillation, high amount of distillation residue

# **Droptube Reactor**

# **System Properties:**

- $p_{svs} = 1$  bar (abs)
- T<sub>max</sub> = up to 1700 °C
- Isothermal zone: 900 mm
- Inner diameter: 70 mmm
- Gas atmosphere: N<sub>2</sub>, CO<sub>2</sub>, Ar, syn. Air
- Gas analytics: FID µGC
- Solid sampling







# Model based description

Numerical model for the evaporation and secondary pyrolysis of a single droplet

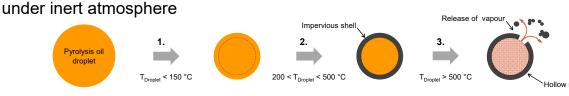


Fig.: Single steps of the conversion of pyrolysis oil; [1]

- 1. Surface Regression
- Evaporation of light volatiles . at liquid surface
- Accumulation of heavy hydrocarbons in surface-near area

### 2. Bubbling/Swelling

- Formation of impervious shell
- Cracking and polymerization of compounds due to temperature increase at outer shell

#### 3. Formation of Solid

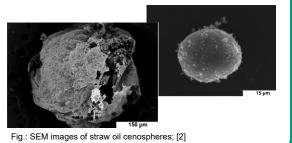
- Solidification of outer shell ٠
  - Release of enclosured vapor creates hollow cenosphere

# Characterization of Cenospheres

Conclusion about the solid formation mechanism and drying of Cenosphere

## Parameters of interest

- Composition and morphology
- Shape and size of cenosphere



[1] Hallett, W.; Clark, N. (2005): A model for the evaporation of biomass pyrolysis oil droplets.

[2] Stoesser, P. et al. (2016): Contribution to the Understanding of Secondary Pyrolysis of Biomass-Based Slurry under Entrained-Flow Gasification Conditions.

KIT – The Research University in the Helmholtz Association